

PREPARATION AND CHARACTERIZATION OF CERAMIC
PORCELAIN PREPARED THROUGH SLIP CASTING

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This report submitted in accordance with requirement of the University Teknikal Malaysia Melaka (UTeM) for the Bachelor Degree of Manufacturing Engineering (Materials Engineering) with Honours.

by

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DECLARATION

I hereby, declared this report entitled “Preparation and Characterization of Ceramic Porcelain Prepared through Slip Casting” is the results of my own research except as cited in references.

Signature :

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Date : 10th April 2009

APPROVAL

This report is submitted to the Faculty of Manufacturing Engineering of UTeM as a partial fulfillment of the requirements for the degree of Bachelor of Manufacturing Engineering (Materials Engineering) with Honours. The member of the supervisory committee is as follow:

(Signature of Supervisor)

.....

(Official Stamp of Supervisor)

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ABSTRACT

Porcelains made of ball clay, feldspar, kaolin and quartz were investigated to study the effect of adjustment of quartz content on the bending strength. These raw materials are mixed by ball milling and formed by slip casting. The bending strength for Batch A is the highest which is 22.51 MPa followed by batch C which is 12.89 MPa and batch B, 8.21MPa. Batch A with the highest quartz content results in lowest reading of total linear shrinkage after drying and sintering which is 24%. A decreasing content of quartz leads to increasing shrinkage. The shrinkage of porcelain increases from bisque firing temperature of 900°C to final sintering temperature of 1250°C. The analysis methods are done to determine the morphologies of the porcelain ceramics using Scanning electron microscopy (SEM) and X-Ray Diffraction (XRD). In XRD analysis, the existence of quartz as SiO₂ is more compared to the existence of mullite in Batch A meanwhile in SEM analysis, the amount of pores in Batch A is less than Batch B and batch C. These analyses have provided the information that increasing in quartz content leads to decreasing in pores amount in a porcelain body and this finding is then correlated to the strength and shrinkage of the body.

ABSTRAK

Porselin diperbuat daripada ball clay, feldspar, kaolin dan quartz telah diselidiki untuk mengetahui kesan perubahan kandungan quartz ke atas kekuatan kelenturannya. Bahan-bahan mentah ini digaul menggunakan penggilingan bebola dan dibentuk dengan teknik penuangan. Kekuatan kelenturan bagi batch A ialah yang tertinggi iaitu 22.51 MPa diikuti oleh batch C dengan 12.89 MPa dan batch B, 8.21MPa. Batch A mempunyai kandungan quartz yang tertinggi menghasilkan bacaan total linear pengecutan selepas pengeringan dan pensinteran yang terendah iaitu 24%. Pengurangan kandungan quartz mengakibatkan peningkatan pengecutan. Pengecutan porselin meningkat dari pembakaran bisque pada 900°C ke pensinteran terakhir pada 1250°C. Analisis yang dilakukan bagi menentukan morfologi porselin menggunakan Scanning electron microscopy (SEM) dan X-Ray Diffraction (XRD). Dalam analisis XRD, kewujudan quartz sebagai SiO₂ lebih berbanding kewujudan mullite dalam Batch A sementara dalam analisis SEM, bilangan pori dalam Batch A kurang daripada Batch B dan Batch C. Analisis –analisis ini telah menyediakan informasi bahawa kandungan quartz memimpin kepada penurunan bilangan liang dalam porselin dan penemuan ini dikaitkan dengan hubungan antara kekuatan dan pengecutan porselin.

DEDICATION

To my beloved mother and family.

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LIST OF ABBREVIATIONS, SYMBOLS & NOMENCLATURE

ACerS	–	American Ceramic Society
ASTM	–	American Society for Testing & Materials
MOR	–	Modulus of Rupture
SEM	–	Scanning Electron Microscopy
SG	–	Specific Gravity
XRD	–	X-Ray Diffraction
$Al_2O_3 \cdot 2SiO_2 \cdot 2H_2O$	–	Ball Clay
$K_2O \cdot Al_2O_3 \cdot 6SiO_2$	–	Feldspar (orthoclase)
$Al_2Si_2O_5(OH)_4$	–	Kaolin
SiO_2	–	Silica or quartz

CHAPTER 1

INTRODUCTION

1.1 Introduction to Ceramics

Ceramics are defined as solids composed of compounds that contain metallic and non metallic elements and the atoms in the compound are held together with strong atomic forces either ionic or covalent bonds (Budinski, 2005). Ceramic engineers and materials engineers are the people who design the processes of fabricating ceramics products. They also create new types of ceramic products and find different uses of ceramic products in everyday life. Ceramics are essential to our daily lifestyle and their applications of products are widely used. Ceramics materials include things like tile, bricks, plates, glass, and toilets. Ceramics can be found in products like watches where it is used as the quartz tuning forks which function as the time keeping devices in watches. Other than that, its application can be found in automobiles where the sparkplugs and engine parts in race cars are made of ceramics. They can also be found on space shuttles and appliances which the enamel coatings are made of ceramics. Ceramics can be dense or lightweight depending on their method of formation. Ceramics usually shows excellent strength and hardness properties. However, they have one significant drawback which is their brittleness but this problem has been solved by the development of new materials such as composites.

Ceramics can be divided into three categories; traditional ceramics or clay based ceramics, advanced ceramics and porous ceramics. Clay based ceramics are produced from minerals mined directly from the earth and falls into three commonly recognized

groups; clay, silica, and feldspar. They are used in a variety of applications including porcelain, refractory and structural clay products. Porous ceramics are developed for their porous natures that are hydrophilic (water loving) and provide capillary wicking and transport of polar substances such as water. The advantages of porous ceramics are that they are not only mechanically strong but inert and can be cleaned and reused in many process applications. The advanced ceramic is the materials that are well defined and their controlled properties are produced from nearly chemically and phase pure starting materials. They can be classified into three categories; oxides ceramics, non-oxides ceramics and composite ceramics. The examples of oxide ceramics are alumina and zirconia. Carbides, borides, nitrides and silicates are in non- oxide ceramics and ceramic based composite are the particulate reinforced and in combinations of oxides and non oxides. The categories of ceramics are as shown in **Figure 1.1**.

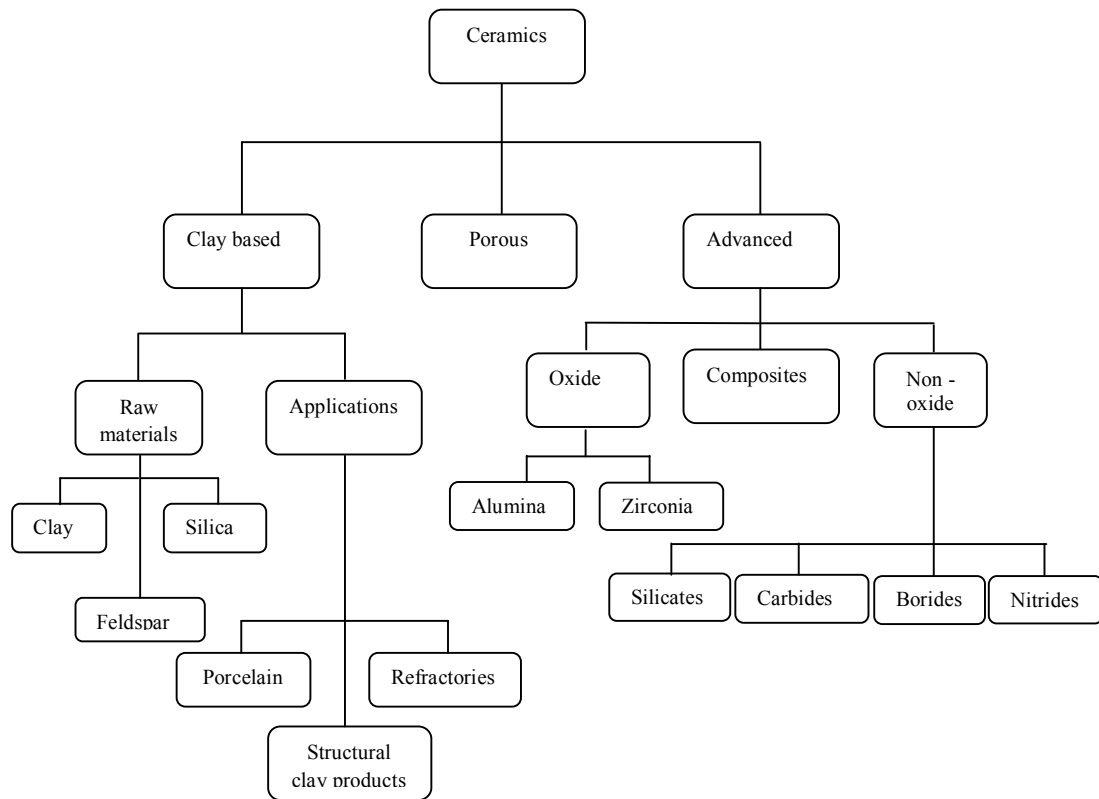


Figure 1.1: Categories of ceramics

The main emphasize in this research is the traditional or clay based ceramics as ceramic porcelain lies in this category. The general classifications of traditional ceramics are as shown in **Table 1.1**.

Table 1.1: The general classifications of traditional ceramics (ACerS, 1920)

Classifications	Descriptions
Pottery	Used as a generic term for ceramics that contains clay and is not used for structural, technical or refractory purposes.
Whiteware	Ceramic ware that is white, ivory or light gray in color after firing. Whiteware is further classified as earthenware, stoneware, chinaware, porcelain and technical ceramics.
Earthenware	Glazed or unglazed non vitreous (porous) clay based ceramic ware. Applications include kitchenware, ovenware, tableware and tile.
Stoneware	Vitreous or semi vitreous ceramic ware of fine texture, made primarily from non refractory fire clay or some combination of clays, fluxes and silica that when fired has property similar to stoneware made from fireclay. Applications include chemicalware, cookware, drainpipe, kitchenware, tableware and tile.
Chinaware	Vitreous ceramic ware of zero or low absorption after firing that is used for non technical applications. Applications include artware, ovenware, sanitaryware, and tableware.
Porcelain	Glazed or unglazed vitreous ceramics ware used primarily for technical purposes. Applications include artware, ball mill balls, chemical ware, insulators and table ware.
Technical ceramics	Vitreous ceramics whiteware used for such products as electrical insulation or for chemical, mechanical, structural or thermal applications.

1.2 Background

The American Society for Testing and Materials (ASTM, 2002) defines a ceramic article as *“an article having a glazed or unglazed body of crystalline or partly crystalline structure, or of glass, which body is produced from essentially inorganic, non-metallic substances and either is formed from a molten mass which solidifies on cooling, or is*

formed and simultaneously or subsequently matured by the action of the heat. The word ceramic is derived from a Greek word 'keramos' meaning pottery. It is related to an old Sanskrit root meaning 'to burn' but primarily, the Greeks used the term to mean "burnt stuff" or "burned earth". Thus the word was used to refer to a product obtained through the action of fire upon earthy materials.

Archeologists have discovered the human-made ceramics that date back to at least 24,000 BC. These ceramics were found in the form of animal and human figurines, slabs, and balls in Czechoslovakia. These ceramics were made of animal fat and bone mixed with bone ash and a fine claylike material. These ceramics were formed and then fired at temperatures in the range of 500 to 800°C in domed and horseshoe shaped kilns partially dug into the ground with loess walls. The function of these ceramics were not clear but is believed not to be the utilitarian one. The first use of functional pottery vessels is thought to be in 9,000 BC or 10,000 BC. These vessels were most likely used to hold and store water, grain and other foods. Tiles were manufactured in Mesopotamia and India almost 10,000 years later as settled communities had been established. Clay bricks were also made around the same time. Glass however was believed to be discovered in Egypt around 8,000 BC, when overheating of kilns produced a colored glaze on the pottery. Experts estimate that glass was produced independently of ceramics and fashioned into separate items before 1,500 BC.

Fast forward to the Middle Ages, the metal industry was in its infancy. At the time, furnaces were constructed of natural materials and were used for melting the metal. The industrial revolution was born when synthetic materials with better resistance to high temperatures called refractories were developed in the 16th century, These refractories created the necessary conditions for melting metals and glass on an industrial scale, as well as for the manufacture of coke, cement, chemicals, and ceramics.

Another major development occurred in the second half of the 19th century, when ceramic materials for electrical insulation were developed (ACerS, 1920). As other

inventions came on the scene including automobiles, radios, televisions and computers, ceramic and glass materials were needed to help these become a reality, as shown in the following timeline in **Table 1.2**.

Table 1.2: Timeline of selected ceramic and glass developments (ACerS, 1920)

Year	Development
24,000 BC	Ceramic figurines used for ceremonial purposes
14,000 BC	First tiles made in Mesopotamia and India
9,000-10,000 BC	Pottery making begins
5,000-8,000 BC	Glazes discovered in Egypt
1,500 BC	Glass objects first made
1550 AD	Synthetic refractories (temperature resistant) for furnaces used to make steel, glass, ceramics, cement
Mid 1800s	Porcelain electrical insulation Incandescent light bulb
1920s	High-strength quartz-enriched porcelain for insulators Alumina sparkplugs Glass windows for automobiles
1930s	-
1940s	Capacitors and magnetic ferrites
1950s	-
1960s	Alumina insulators for voltages over 220 kV Application of carbides and nitrides
1970s	Introduction of high-performance cellular ceramic substrates for catalytic converters and particulate filters for diesel engines
1980s	High temperature superconductors

1.3 Problem Statement

Many forming processes used in ceramics such as dry pressing, extrusion, plastic forming and injection moulding are similar in principle of processing plastics and metals. However, the abrasiveness, wear resistance and hardness of ceramics are the difficulties faced in ceramic processing in order to avoid wear and contamination. Due to that problem, slip casting is chosen as a more specifically forming method used for ceramic processing. Good material homogeneity can also be achieved in slip casting. Furthermore, slip casting is a conventional method and the inexpensive mould material makes slip casting as a low cost fabrication method. In this research, ceramics to be fabricated is the porcelain ceramics with varying proportions of quartz and kaolin and it is sintered as high as 1250°C. Sintering temperature causes densification that gives the ceramic products strength and other properties that will be characterized in this research.

1.4 Objectives

The objectives of this research are:

- a) To produce high density sintered porcelain at sintering temperature of 1250°C.
- b) To identify the optimum composition of quartz and kaolin in yielding the highest bending strength of a porcelain body.
- c) To characterize and determine the mechanical strength of final product by varying the proportions of quartz and kaolin.

1.5 Scope

In this research, the study will be focused on the ceramic processing technique as a method to get high density porcelain ceramics when sintered at 1250°C. The kaolin and quartz proportions will be varied while other raw materials like feldspar, ball clay and

alumina will be kept constant. These raw materials will be mixed together using wet ball milling and cast into solid. The effects of varying proportions of kaolin and quartz on porcelain strength will be tested by drying and shrinkage test, flexural test, hardness test and tensile test. The morphology of the sintered porcelain body will be observed using SEM and XRD. The information getting from conducting the test will be compared so that the optimum compositions of kaolin and quartz required to get highest strength of the porcelain body can be achieved. Chapter One introduces the ceramics generally and its background history. Chapter Two describes mainly the porcelain and the previous research related to this research and Chapter Three describes the method of sample preparation and evaluation of final product. Chapter Four states the result and discussions and Chapter Five concludes the present research.